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$\nu_1 + \nu_3$ Combination Band of SO_2

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December 1972

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THE UNIVERSITY OF TENNESSEE
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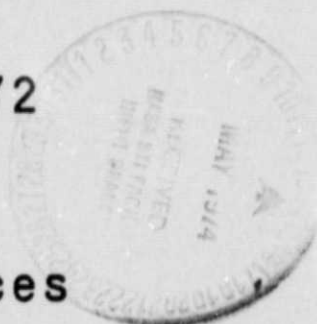
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$\nu_1 + \nu_3$ COMBINATION BAND OF $^{32}\text{S}^{16}\text{O}_2$

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ABSTRACT

The infrared-active vibration-rotation combination band $\nu_1 + \nu_3$ of sulfur dioxide has been measured with moderately high spectral resolution. Quantum number identifications of spectral lines were made by comparison with theoretically computed spectra which include the effects of centrifugal distortion. Relative line intensities were also calculated. The band center for $\nu_1 + \nu_3$ has been determined to be $2499.60 \pm 0.10 \text{ cm}^{-1}$.

I. INTRODUCTION

Recently we investigated the infrared-active vibration-rotation fundamental bands of sulfur dioxide at moderately high resolution.^{1,2} We felt that earlier results for some combination and overtone bands obtained by Shelton, Nielsen, and Fletcher³ could also be improved considerably, and that these bands could be analyzed by comparison with theoretically computed spectra. Consequently, as part of this program, we have obtained spectra of the $\nu_1 + \nu_3$ combination band of $^{32}\text{S}^{16}\text{O}_2$ at a resolution of about 0.15 cm^{-1} . We have assigned quantum numbers to the observed lines by using theoretical spectra which include the effects of centrifugal distortion. All transitions with $J \leq 60$ were included. Relative line intensities of these transitions were also calculated.

Experimental details of our work are given in the next Section. In Section III, the theoretical procedure is discussed briefly. In the last Section, we present our results in the form of measured spectra, and tabulated experimental and theoretical line positions. Possible applications of our results to atmospheric problems are considered.

II. EXPERIMENTAL DETAILS

The anhydrous grade $^{32}\text{S}^{16}\text{O}_2$ gas sample was obtained from Matheson Gas Products. The stated purity of the sample was 99.98% by weight; with 50, 10, and 30 p.p.m. maxima of H_2O , H_2SO_4 , and non-volatiles, resp. A Perkin-Elmer Model 225 Grating Infrared Spectrophotometer, equipped with an f/5 fore monochromator for pre-dispersion of the radiation, was used to record the spectra. A 150 lines/mm grating was employed in the second order for the 2465 to 2525 cm^{-1} region measured. A servo-controlled slit

program was utilized to provide constant energy to the thermopile detector. All scans were recorded on a Model 225 Auxiliary Recorder with a dispersion of 1 cm/cm^{-1} and at a speed of $0.5 \text{ cm}^{-1}/\text{min}$. Of the six runs taken under identical conditions, the best two with respect to reproducibility, resolution, and noise level were selected for the theoretical analysis of the band.

A 10-cm absorption cell, equipped with KCl windows and containing 50 torr of sulfur dioxide, was used to record the spectra. The cell was placed in the sample compartment for one hour prior to each run in order to stabilize (at approximately 315°K) the temperature increase caused by heating from the Globar radiation source. Slit widths were set at $0.22 \pm 0.04 \text{ cm}^{-1}$. The ν_2 band of C_2D_2 was used for calibration purposes.⁴

III. THEORETICAL PROCEDURE

The theoretical approach utilized here is the same as that used by us for the analysis of the fundamental bands of this molecule. Details are given in Refs. 1 and 2. The rotational constants of the ground state, and the centrifugal distortion constants (assumed to be the same in the ground and excited states) were given in Table II of Refs. 1 and 2. The values of the excited-state rotational constants $A = 2.007487$, $B = 0.341346$, and $C = 0.290987 \text{ cm}^{-1}$, were taken from the microwave work of Saito⁵ on pure rotational transitions from $\nu_1 + \nu_3$ of sulfur dioxide.

The computer program, described in Refs. 1 and 2, takes into account the fact that in the ground state of SO_2 only symmetric levels occur because of C_{2v} symmetry and zero spin of the ^{16}O nuclei. The initial value for the $\nu_1 + \nu_3$ band center was taken from the work of Shelton et al.³ This value was later adjusted so that the experimental and theoretical spectra would match.

IV. RESULTS AND DISCUSSION

The results of our experimental and theoretical studies of the $\nu_1 + \nu_3$ band of $^{32}\text{S}^{16}\text{O}_2$ are given in Fig. 1 and Table I. A representative portion of these results, as well as band intensity and dipole moment derivative data, are presented in Ref. 6. The spectrum, with resolution of about 0.15 cm^{-1} , extends from 2465 to 2525 cm^{-1} . In Figs. 1(a) and 1(b), the upper tracings represent measured experimental spectra, with percent absorption shown in the right-hand vertical scales. The lower tracings are calculated theoretical spectra. Their relative intensities, referred to the left-hand vertical scales, are normalized distributions of line intensities.^{1,2} As can be seen in the Figure, there is good agreement between the experimental and theoretical spectra.

In Table I, the observed and calculated spectral line positions are compared. The absolute accuracy of measured line positions is $\pm 0.10 \text{ cm}^{-1}$. We estimate the relative accuracy to be $\pm 0.05 \text{ cm}^{-1}$. Only the stronger theoretical transitions corresponding to each experimental line peak have been listed in the Table. (Where there is a blank space in the column of experimental line positions, the adjacent theoretical line position corresponds to the previously tabulated experimental line.) Initial- and final-state quantum numbers, as well as theoretical relative line intensities $I_{n''n'}^0/C'$ defined by Eq. (6) in Ref. 1 [Eq. (4) in Ref. 2], are given for each theoretical transition. The quantum numbers K_{-1} and K_1 are associated with the projection of the total angular momentum (having quantum number J) on the symmetry axis in, resp., the prolate and oblate symmetric top limiting cases.^{7,8}

$\nu_1 + \nu_3$, like the ν_3 fundamental, is a type A band. It has a strong central Q branch. The selection rules⁹ are $\Delta J = 0, \pm 1$; $J = 0 \rightarrow 0$;

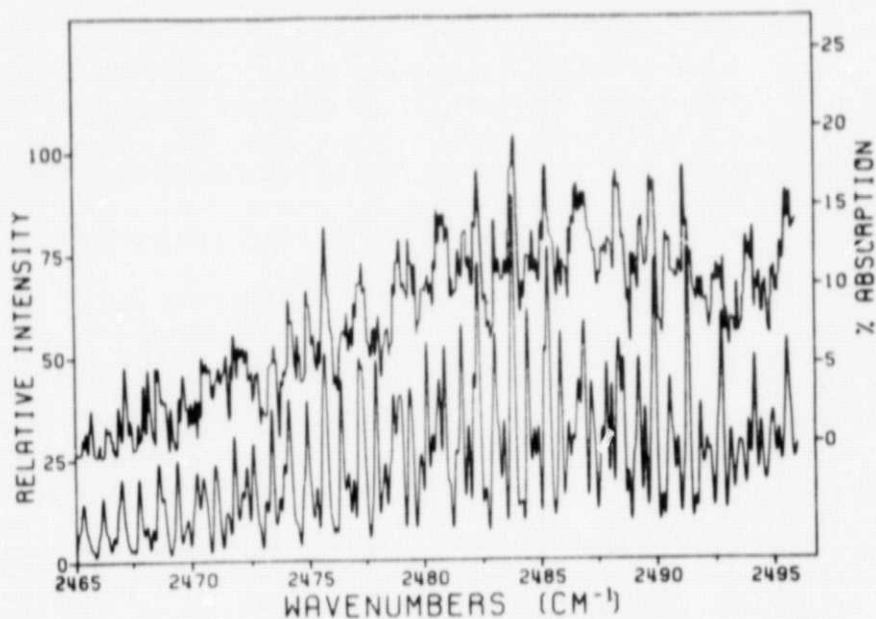


Fig. 1(a). Experimental and theoretical spectra of $\nu_1 + \nu_3$ band of $^{32}\text{S}^{16}\text{O}_2$ in range 2465 to 2495 cm^{-1} .

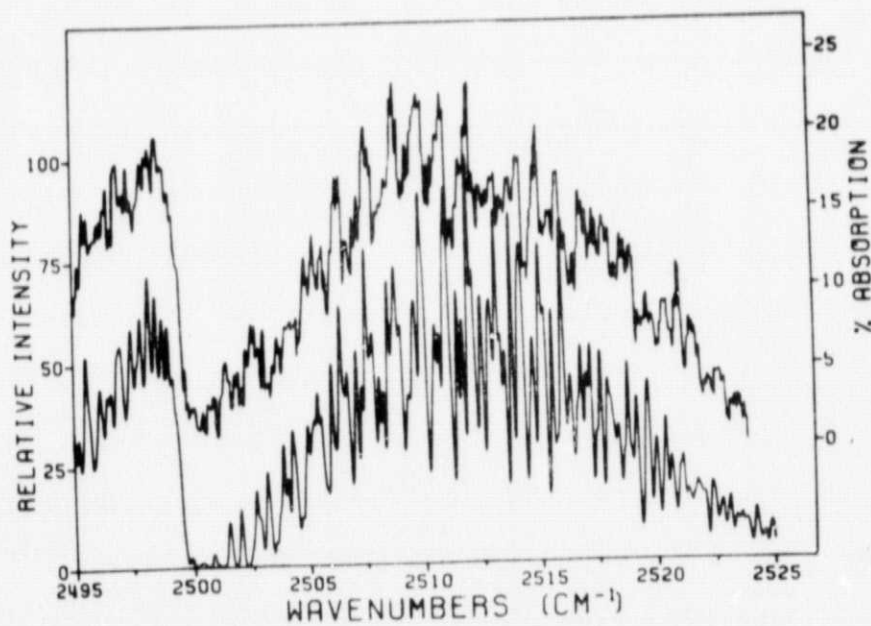


Fig. 1(b). Experimental and theoretical spectra of $\nu_1 + \nu_3$ band of $^{32}\text{S}^{16}\text{O}_2$ in range 2495 to 2525 cm^{-1} .

TABLE I

COMPARISON OF EXPERIMENTAL AND THEORETICAL SPECTRAL LINE POSITIONS, WITH QUANTUM
NUMBER ASSIGNMENTS, FOR THE $\nu_1 + \nu_3$ BAND OF $^{32}\text{S}^{16}\text{O}_2$ CENTERED AT $2499.60 \pm 0.10 \text{ cm}^{-1}$.
LINE INTENSITIES [SEE EQ. (6) OF REF. 1, OR EQ. (4) OF REF. 2] ARE COMPUTED AT 300°K .

Line Position (in cm^{-1})		Quantum Numbers							Line Position (in cm^{-1})		Quantum Numbers						
Exptl.	Theor.	J'	K'_{-1}	K'_1	J''	K''_{-1}	K''_1	$\frac{I^\circ_{n''n'}}{G^\circ}$	Exptl.	Theor.	J'	K'_{-1}	K'_1	J''	K''_{-1}	K''_1	$\frac{I^\circ_{n''n'}}{G^\circ}$
2465.34	2465.31	43	7	36	44	7	37	1.3624		2471.48	33	14	19	34	14	20	0.9166
	2465.33	47	0	47	48	0	48	1.6121		2471.50	31	17	14	32	17	50	0.4275
	2465.37	46	2	45	47	2	46	1.6130	2471.68	2471.69	30	18	13	31	18	14	0.3167
2465.49	2465.49	42	10	33	43	10	34	0.9843	2471.81	2471.77	36	5	32	37	5	33	3.3916
2465.74	2465.79	42	9	34	43	9	35	1.1603		2471.77	35	9	26	36	9	27	2.2546
2466.12	2466.08	40	13	28	41	13	29	0.6604		2471.80	39	0	39	40	0	40	3.7538
	2466.16	46	1	46	47	1	47	1.8353		2471.83	37	2	35	38	2	36	3.6731
2466.39	2466.36	41	10	31	42	10	32	1.0943		2471.83	38	2	37	39	2	38	3.6863
	2466.40	42	6	37	43	6	38	1.7002	2472.02	2472.04	35	8	27	36	8	28	2.6242
2466.53	2466.50	40	12	29	41	12	30	0.8233	2472.20	2472.16	35	4	31	36	4	32	3.9094
	2466.55	37	17	20	38	17	21	0.2946		2472.25	35	7	28	36	7	29	2.9973
2466.92	2466.88	40	11	30	41	11	31	1.0080	2472.35	2472.31	35	5	30	36	5	31	3.6729
	2466.89	41	5	36	42	5	37	2.0410		2472.37	33	12	21	34	12	22	1.4786
	2466.91	41	8	33	42	8	34	1.4951		2472.38	35	6	29	36	6	30	3.3566
	2466.93	37	13	26	40	13	27	0.7265	2472.53	2472.55	35	3	32	36	3	33	4.0961
2467.21	2467.20	42	4	39	43	4	40	2.0436		2472.58	38	1	38	39	1	39	4.1019
	2467.23	40	10	31	41	10	32	1.2120	2472.73	2472.69	36	3	34	37	3	35	4.0005
2467.40	2467.36	39	12	27	40	12	28	0.9067		2472.76	33	11	22	34	11	23	1.8250
	2467.39	36	17	20	37	17	21	0.3178		2472.76	32	13	20	33	13	21	1.2528
2467.78	2467.78	38	13	26	39	13	27	0.7959	2473.47	2473.43	34	5	30	35	5	31	4.0126
	2467.79	40	8	33	41	8	34	1.6583		2473.43	33	9	24	34	9	25	2.6272
2468.00	2467.98	40	7	34	41	7	35	1.8862	2473.78	2473.74	34	4	31	35	4	32	4.3558
2468.22	2468.20	38	12	27	39	12	28	0.9943	2473.93	2473.91	30	14	17	31	14	18	1.0850
	2468.22	35	17	18	36	17	19	0.3412		2473.92	33	7	26	34	7	27	3.5069
	2468.25	5	3	2	6	5	1	1.4268		2473.94	32	10	23	33	10	24	2.3675
2468.51	2468.50	36	15	22	37	15	23	0.5685	2474.12	2474.08	33	6	27	34	6	28	3.9358
2468.68	2468.64	42	2	41	43	2	42	2.5049		2474.11	33	3	30	34	3	31	4.8343
	2468.64	39	8	31	40	8	32	1.8322		2474.14	36	1	36	37	1	37	4.8469
	2468.71	39	5	34	40	5	35	2.5235		2474.16	35	1	34	36	1	35	4.7498
	2468.73	39	4	35	40	4	36	2.6688	2474.27	2474.25	32	9	24	33	9	25	2.8175
2468.81	2468.80	35	16	19	36	16	20	0.4616		2474.26	34	3	32	35	3	33	4.7111
2468.96	2468.94	38	10	29	39	10	30	1.4690	2474.50	2474.52	32	8	25	33	8	26	3.2901
2469.19	2469.24	38	9	30	39	9	31	1.7371	2474.90	2474.90	35	0	35	36	0	36	5.2408
2469.51	2469.51	40	3	38	41	3	39	2.7638		2474.90	33	2	31	34	2	32	5.1070
	2469.52	38	8	31	39	8	32	2.0165		2474.93	32	6	27	33	6	28	4.2363
2469.70	2469.71	38	7	32	39	7	33	2.2973		2474.94	34	2	33	35	2	34	5.1306
2470.05	2470.08	38	5	34	39	5	35	2.8223	2475.22	2475.18	29	13	16	30	13	17	1.4695
	2470.09	37	9	28	38	9	29	1.9026		2475.20	30	11	20	31	11	21	2.2009
2470.21	2470.21	41	0	41	42	0	42	3.1116	2475.48	2475.47	26	17	10	27	17	11	0.4801
	2470.24	40	2	39	41	2	40	3.0595		2475.51	28	14	15	29	14	16	1.1788
2470.36	2470.35	37	8	29	38	8	30	2.2104	2475.73	2475.70	31	3	28	32	3	29	5.6191
2470.50	2470.49	38	4	35	39	4	36	3.0713		2475.70	33	1	32	34	1	33	5.5245
	2470.53	37	5	32	38	5	33	3.0697		2475.74	31	6	25	32	6	26	4.5377
2470.64	2470.63	36	10	27	37	10	28	1.7519	2476.17	2476.15	30	8	23	31	8	24	3.7430
	2470.66	37	6	31	38	6	32	2.8159	2476.49	2476.47	32	2	31	33	2	32	5.9230
2470.90	2470.89	31	18	13	32	18	14	0.3057	2476.70	2476.68	29	9	20	30	9	21	3.3815
	2470.93	36	9	28	37	9	29	2.0754		2476.72	30	5	26	31	5	27	5.3483
2471.18	2471.13	34	13	22	35	13	23	1.0973	2476.90	2476.93	30	4	27	31	4	28	5.8071
	2471.20	36	8	29	37	8	30	2.4134	2477.06	2477.08	26	14	13	27	14	14	1.2458
2471.48	2471.46	35	10	25	36	10	26	1.9012	2477.21	2477.19	32	1	32	33	1	33	6.4829

TABLE I (Continued)

Line Position (in cm^{-1})		Quantum Numbers							Line Position (in cm^{-1})		Quantum Numbers						
Exptl.	Theor.	J'	K' ₋₁	K' ₁	J''	K'' ₋₁	K'' ₁	$\frac{I^2 n'' n'}{C'}$	Exptl.	Theor.	J'	K' ₋₁	K' ₁	J''	K'' ₋₁	K'' ₁	$\frac{I^2 n'' n'}{C'}$
	2477.19	29	7	22	30	7	23	4.5595	2484.50	2484.48	22	1	22	23	1	23	10.2136
	2477.21	31	1	30	32	1	31	6.3317	2484.71	2484.70	19	8	11	20	8	12	5.2278
2477.36	2477.36	30	3	28	31	3	29	6.2463		2484.71	20	4	17	21	4	18	8.7312
	2477.37	29	6	23	30	6	24	5.1412	2484.89	2484.90	20	3	18	21	3	19	9.3999
2477.50	2477.48	28	9	20	29	9	21	3.5592	2485.12	2485.10	16	13	4	17	13	5	1.1068
	2477.55	26	13	14	27	13	15	1.6329		2485.15	18	9	10	19	9	11	4.2422
2477.91	2477.88	29	2	27	30	2	28	6.7047		2485.16	21	0	21	22	0	22	10.4392
	2477.94	31	0	31	32	0	32	6.9079		2485.16	19	6	13	20	6	14	7.1202
	2477.95	27	10	17	28	10	18	3.1048	2485.29	2485.28	20	2	19	21	2	20	9.9873
2478.10	2478.13	24	15	10	25	15	11	0.9366		2485.32	19	2	17	20	2	18	9.9040
2478.51	2478.52	28	4	25	29	4	26	6.5410		2485.33	19	5	14	20	5	15	8.0229
	2478.55	27	8	19	28	8	20	4.3857		2485.34	19	3	16	20	3	17	9.4804
2478.70	2478.69	30	1	30	31	1	31	7.3328	2485.85	2485.80	19	1	18	20	1	19	10.2192
	2478.69	29	1	28	30	1	29	7.1478		2485.87	20	1	20	21	1	21	10.6112
	2478.73	26	10	17	27	10	18	3.2263		2485.89	18	6	13	19	6	14	7.1157
2478.84	2478.80	27	7	20	28	7	21	5.0588	2486.06	2486.08	18	5	14	19	5	15	8.0469
	2478.89	28	3	26	29	3	27	7.0244	2486.21	2486.18	17	8	9	18	8	17	5.0655
2479.00	2478.98	27	6	21	28	6	22	5.7200		2486.22	18	4	15	19	4	16	8.8884
	2479.01	27	4	23	28	4	24	6.8560	2486.40	2486.37	18	3	16	19	3	17	9.6017
2479.18	2479.16	25	11	14	26	11	15	2.7043		2486.43	17	7	10	18	7	11	6.0576
2479.40	2479.40	23	14	9	24	14	10	1.2688	2486.55	2486.55	19	0	19	20	0	20	10.7400
	2479.43	29	0	29	30	0	30	7.7540	2486.69	2486.70	18	2	17	19	2	18	10.2017
2479.64	2479.59	26	7	20	27	7	21	5.2898	2486.82	2486.82	17	2	15	18	2	16	10.0660
	2479.67	22	15	8	23	15	9	0.8989		2486.82	17	5	12	18	5	13	8.0129
2480.19	2480.16	28	1	28	29	1	29	8.1668	2486.96	2486.94	17	4	13	18	4	14	8.8784
	2480.16	27	1	26	28	1	27	7.9441	2487.39	2487.38	16	6	11	17	6	12	6.9352
2480.37	2480.37	25	7	18	26	7	19	5.5033	2487.58	2487.56	16	5	12	17	5	13	7.9170
	2480.39	26	3	24	27	3	25	7.7682	2487.71	2487.70	16	4	13	17	4	14	8.8090
2480.51	2480.53	25	3	22	26	3	23	7.9730	2487.86	2487.83	16	3	14	17	3	15	9.5646
2480.62	2480.61	24	9	16	25	9	17	4.1414		2487.89	15	7	8	16	7	9	5.7105
2480.80	2480.78	26	6	21	27	6	22	5.9908		2487.90	17	0	17	18	0	18	10.8167
	2480.84	25	9	16	26	9	17	4.0217	2488.06	2488.06	14	9	6	15	9	7	3.4243
2481.10	2481.06	23	10	13	24	10	14	3.4886		2488.10	16	2	15	17	2	16	10.1782
	2481.14	24	7	18	25	7	19	5.6950	2488.33	2488.33	15	2	13	16	2	14	9.9654
2481.38	2481.35	24	6	19	25	6	20	6.4733		2488.35	14	8	7	15	8	8	4.4055
	2481.38	23	9	14	24	9	15	4.2382	2488.45	2488.42	15	4	11	16	4	12	8.6690
	2481.39	21	13	18	22	13	19	1.6530		2488.44	15	3	12	16	3	13	9.4307
2481.51	2481.50	24	5	20	25	5	21	7.2075	2488.63	2488.61	14	7	8	15	7	9	5.4392
2481.80	2481.83	25	2	23	26	2	24	8.2761		2488.62	16	1	16	17	1	17	10.7474
2481.96	2481.92	23	7	16	24	7	17	5.8605	2488.77	2488.78	13	9	4	14	9	5	3.0677
	2481.92	19	15	4	20	15	5	0.7283	2489.02	2489.01	14	5	10	15	5	11	7.5220
2482.15	2482.12	23	6	17	24	6	18	6.6760		2489.07	13	8	5	14	8	6	4.0473
	2482.14	23	3	20	24	3	21	8.6297	2489.15	2489.14	14	4	11	15	4	12	10.4250
	2482.15	22	9	14	23	9	15	4.3081		2489.16	12	10	3	13	10	4	1.7708
2482.34	2482.32	23	2	21	24	2	22	8.9618	2489.35	2489.33	13	7	6	14	7	7	5.0964
	2482.34	25	0	25	26	0	26	9.3131	2489.76	2489.73	13	5	8	14	5	9	7.2152
2482.58	2482.57	20	12	9	21	12	10	2.1435		2489.78	12	8	5	13	8	6	3.6190
	2482.58	21	10	11	22	10	12	3.5458	2489.92	2489.87	13	4	9	14	4	10	8.1738
2482.88	2482.89	22	6	17	23	6	18	6.8470		2489.93	13	3	10	14	3	11	8.9832
	2482.71	21	9	12	22	9	13	4.3475		2489.96	14	1	14	15	1	15	10.4250
2483.03	2483.01	23	1	22	24	1	23	9.3397		2489.96	13	1	12	14	1	13	9.8935
	2483.05	22	5	18	23	5	19	7.6532	2490.43	2490.44	12	5	8	13	5	9	6.8304
2483.20	2483.19	22	4	19	23	4	20	8.3760	2490.59	2490.56	13	0	13	14	0	14	10.1894
	2483.19	21	8	13	22	8	14	5.2070		2490.59	12	4	9	13	4	10	7.8111
2483.47	2483.44	21	7	14	22	7	15	6.0965	2490.96	2490.96	11	6	5	12	6	6	5.2780
2483.91	2483.88	21	4	17	22	4	18	8.5664	2491.15	2491.15	11	5	6	12	5	7	6.3635
	2483.95	20	8	13	21	8	14	5.2382		2491.18	10	8	3	11	8	4	2.5065
2484.24	2484.21	20	7	14	21	7	15	6.1573	2491.28	2491.29	11	4	7	12	4	8	7.3674

TABLE I (Continued)

Line Position (in cm^{-1})		Quantum Numbers							Line Position (in cm^{-1})		Quantum Numbers						
Exptl.	Theor.	J'	K' ₁	K' ₂	J''	K'' ₁	K'' ₂	$\frac{I''_{n''n'}}{G'}$	Exptl.	Theor.	J'	K' ₁	K' ₂	J''	K'' ₁	K'' ₂	$\frac{I''_{n''n'}}{G'}$
	2491.30	12	1	12	13	1	13	9.8156		2496.29	4	2	3	5	2	4	3.8820
2491.52	2491.49	21	20	1	21	20	2	0.6928		2496.30	17	12	5	17	12	6	3.1687
	2491.51	26	19	8	26	19	7	0.4835		2496.32	26	9	18	26	9	17	1.0701
	2491.54	30	18	13	30	18	12	0.3600	2496.43	2496.40	16	12	5	16	12	4	3.5409
2491.81	2491.79	24	19	6	24	19	5	0.6113		2496.42	29	7	22	29	7	23	0.5750
	2491.80	32	17	16	32	17	15	0.3308		2496.46	25	9	16	25	9	17	1.2051
	2491.82	35	16	19	35	16	20	0.2570		2496.47	4	1	4	5	1	5	4.5549
	2491.85	10	5	6	11	5	7	5.8101	2496.56	2496.56	14	12	3	14	12	2	4.4319
2491.99	2492.00	10	4	7	11	4	8	6.8403	2496.72	2496.70	17	11	6	17	11	7	3.2144
2492.13	2492.11	10	3	8	11	3	9	7.7255		2496.71	12	12	1	12	12	0	5.5864
	2492.13	9	7	2	10	7	3	2.8850		2496.73	23	9	14	23	9	15	1.5214
2492.41	2492.38	19	19	0	19	9	1	1.0759		2496.75	20	10	11	20	10	10	2.2600
	2492.42	24	18	7	24	18	6	0.7429	2496.95	2496.94	3	2	1	4	2	2	2.8155
2492.58	2492.54	9	5	4	10	5	5	5.1642		2496.96	14	11	4	14	11	3	4.4959
	2492.59	10	1	10	11	1	11	8.9121		2496.96	27	6	21	27	6	22	0.5952
2492.71	2492.69	9	4	5	10	4	6	6.2250		2496.97	21	9	12	21	9	13	1.9105
2492.86	2492.85	35	14	21	35	14	22	0.3213	2497.26	2497.24	15	10	5	15	10	6	3.9420
	2492.87	29	16	13	29	16	14	0.5625		2497.26	21	8	13	21	8	14	1.7329
	2492.88	25	17	8	25	17	9	0.7855		2497.29	18	9	10	18	9	9	2.6711
2493.02	2493.01	19	18	1	19	18	2	1.3075	2497.47	2497.46	12	10	3	12	10	2	5.5628
	2493.02	24	17	8	24	17	7	0.8827		2497.48	16	9	8	16	9	7	3.3356
	2493.04	8	6	3	9	6	4	3.2440		2497.48	19	8	11	19	8	12	2.1688
2493.16	2493.15	9	0	9	10	0	10	8.4390	2497.61	2497.58	18	8	11	18	8	10	2.4242
2493.31	2493.28	22	17	6	22	17	5	1.1100		2497.59	10	10	1	10	10	0	7.1106
	2493.31	35	13	22	35	13	23	0.3453		2497.60	2	2	1	3	2	2	1.5866
	2493.32	26	16	11	26	16	10	0.8102		2497.62	22	6	17	22	6	16	1.0848
2493.48	2493.51	8	3	6	9	3	7	6.4742		2497.63	20	7	14	20	7	13	1.6757
2493.81	2493.81	17	17	0	17	17	1	1.9391		2497.64	14	9	6	14	9	5	4.1758
	2493.84	22	16	7	22	16	6	1.2885	2498.03	2498.01	13	8	5	13	8	6	4.2522
	2493.85	26	15	12	26	15	11	0.9178		2498.03	16	7	10	16	7	9	2.6194
2493.97	2493.96	21	16	5	21	16	6	1.4424		2498.04	20	5	16	20	5	15	1.0263
	2493.99	25	15	10	25	15	11	1.0329	2498.24	2498.20	10	8	3	10	8	2	6.1113
2494.12	2494.12	24	15	10	24	15	9	1.1607		2498.25	16	6	11	16	6	10	2.1371
	2494.15	7	1	6	8	1	7	6.9629		2498.25	18	5	14	18	5	13	1.2899
2494.24	2494.25	23	15	8	23	15	9	1.3025		2498.25	1	1	0	2	1	1	1.4736
	2494.28	18	16	3	18	16	2	2.0148		2498.26	9	8	1	9	8	2	6.9684
2494.39	2494.36	29	13	16	29	13	17	0.7566		2498.27	13	7	6	13	7	7	3.6799
	2494.38	17	16	1	17	16	2	2.2508	2498.53	2498.34	12	7	6	12	7	5	4.1376
	2494.38	22	15	8	22	15	7	1.4598		2498.34	15	6	9	15	6	10	2.3914
	2494.40	6	6	1	7	6	2	1.2693		2498.40	11	7	4	11	7	5	4.6678
2494.63	2494.61	20	15	6	20	15	5	1.8279	2498.63	2498.61	7	7	0	7	7	1	8.0630
	2494.63	24	14	11	24	14	10	1.2821		2498.61	14	5	10	14	5	9	2.0307
	2494.63	30	12	19	31	12	18	0.6973		2498.63	11	6	5	11	6	6	3.8133
2495.02	2495.01	16	15	2	16	15	1	2.8496	2498.87	2498.87	6	6	1	6	6	0	7.7800
	2495.02	30	11	20	30	11	19	0.7066		2498.87	10	5	6	10	5	5	3.2825
2495.23	2495.22	6	1	6	7	1	7	6.2626		2498.70	12	4	9	12	4	8	1.7604
	2495.22	19	14	5	19	14	6	2.2570	2499.11	2499.09	5	5	0	5	5	1	7.1336
2495.52	2495.49	27	11	16	27	11	17	1.0244		2499.09	9	4	5	9	4	6	2.5764
	2495.50	16	14	3	16	14	2	3.1479		2499.13	8	4	5	8	4	4	2.9642
	2495.51	24	12	13	24	12	12	1.4414	2499.83	2499.82	21	3	18	21	3	19	0.3146
	2495.53	5	1	4	6	1	5	5.4105		2499.85	3	1	2	3	1	3	0.5686
	2495.53	5	3	2	6	3	3	3.9199	2500.20	2500.22	1	0	1	0	0	0	0.9995
2495.71	2495.68	19	13	6	19	13	7	2.4277	2500.47	2500.45	23	3	20	23	3	21	0.2355
	2495.73	5	0	5	6	0	6	5.6186	2500.78	2500.79	2	1	2	1	1	1	1.4834
2496.13	2496.11	19	12	7	19	12	8	2.5385	2500.99	2500.98	15	2	13	15	2	14	0.2904
	2496.13	14	13	2	14	13	1	4.2381	2501.44	2501.42	3	2	1	2	2	0	1.5982
	2496.14	25	10	15	25	10	16	1.2753		2501.48	3	0	3	2	0	2	2.9713
2496.29	2496.28	21	11	10	21	11	11	2.0589	2501.99	2501.98	4	1	4	3	1	3	3.6540

TABLE I (Continued)

Line Position (in cm^{-1})		Quantum Numbers							Line Position (in cm^{-1})		Quantum Numbers						
Exptl.	Theor.	J'	K' ₋₁	K' ₁	J''	K'' ₋₁	K'' ₁	$\frac{I''n''n'}{G'}$	Exptl.	Theor.	J'	K' ₋₁	K' ₁	J''	K'' ₋₁	K'' ₁	$\frac{I''n''n'}{G'}$
	2502.03	4	2	3	3	2	2	2.8504		2509.23	17	5	12	16	5	11	8.3407
2502.44	2502.44	5	4	1	4	4	0	1.5313	2509.40	2509.41	17	4	13	16	4	12	9.2800
2502.70	2502.69	5	0	5	4	0	4	4.8434	2509.91	2509.89	19	7	12	18	7	11	6.5107
2502.85	2502.81	5	1	4	4	1	3	4.6108		2509.92	18	2	17	17	2	16	10.7908
	2502.89	6	5	2	5	5	1	1.4268		2509.95	18	4	15	17	4	14	9.3853
2503.06	2503.04	6	4	3	5	4	2	2.7927	2510.09	2510.06	19	0	19	18	0	18	11.4181
2503.27	2503.24	6	2	5	5	2	4	4.9292		2510.12	19	6	13	18	6	12	7.5425
	2503.30	7	6	1	6	6	0	1.2968	2510.32	2510.32	19	5	14	18	5	13	8.5299
2503.47	2503.49	7	5	2	6	5	1	2.6198		2510.35	23	12	11	22	12	10	2.3716
2503.63	2503.64	7	4	3	6	4	2	3.8778	2510.50	2510.51	20	1	20	19	1	19	11.3710
	2503.67	8	7	2	7	7	1	1.1520		2510.52	19	4	15	18	4	14	9.4201
2503.81	2503.77	7	3	4	6	3	3	4.9776	2510.69	2510.69	19	1	18	18	1	17	10.9375
	2503.85	7	0	7	6	0	6	6.5507		2510.69	21	8	13	20	8	12	5.5863
2504.02	2504.01	9	8	1	8	8	0	1.0015	2510.90	2510.91	22	9	14	21	9	13	4.6506
	2504.05	7	1	6	6	1	5	6.3590		2510.93	20	2	19	19	2	18	10.7478
2504.35	2504.37	8	3	6	7	3	5	5.8615		2510.95	21	7	14	20	7	13	6.5688
2504.50	2504.49	9	6	3	8	6	2	3.3346	2511.06	2511.03	19	2	17	18	2	16	10.6336
2504.94	2504.97	9	0	9	8	0	8	8.0409		2511.03	20	4	17	19	4	16	9.3973
	2504.97	9	3	6	8	3	5	6.6500		2511.10	20	3	18	19	3	17	10.1301
2505.09	2505.07	10	6	5	9	6	4	4.1523		2511.10	23	10	13	22	10	12	3.7890
	2505.13	9	2	7	8	2	6	7.4081	2511.19	2511.18	21	6	15	20	6	14	7.5445
	2505.13	12	10	3	11	10	2	1.3026	2511.66	2511.63	21	1	20	20	1	19	10.7251
2505.29	2505.26	10	5	6	9	5	5	5.3248		2511.63	21	4	17	20	4	16	9.3077
	2505.27	9	1	8	8	1	7	7.8521		2511.71	22	6	17	21	6	16	7.4687
2505.71	2505.70	13	10	3	12	10	2	1.8427	2511.90	2511.90	22	2	21	21	2	20	10.4689
	2505.75	12	8	5	11	8	4	3.2247		2511.92	22	5	18	21	5	17	8.3682
2506.04	2506.00	11	4	7	10	4	6	7.0748	2512.09	2512.09	21	2	19	20	2	18	10.4020
	2506.00	12	7	6	11	7	5	4.3299		2512.09	22	4	19	21	4	18	9.1760
	2506.02	13	9	4	12	9	3	2.7459		2512.13	22	3	20	21	3	19	9.8632
	2506.05	11	0	11	10	0	10	9.2781	2512.23	2512.23	23	6	17	22	6	16	7.3478
2506.35	2506.32	13	8	5	12	8	4	3.7659		2512.24	24	8	17	23	8	16	5.4221
	2506.36	11	2	9	10	2	8	8.6816	2512.52	2512.51	24	7	18	23	7	17	6.3085
2506.52	2506.46	12	1	12	11	1	11	9.7348		2512.55	23	1	22	22	1	21	10.3083
	2506.57	13	7	6	12	7	5	4.8865	2512.70	2512.72	23	4	19	22	4	18	8.9790
2506.73	2506.71	12	2	11	11	2	10	9.2244	2512.84	2512.85	24	2	23	2	2	22	9.9929
	2506.71	12	3	10	11	3	9	8.5354		2512.86	25	0	25	24	0	24	10.3656
2507.10	2507.09	13	0	13	12	0	12	10.2402	2513.00	2512.97	24	5	20	23	5	19	8.0171
	2507.14	14	7	8	13	7	7	5.3196		2513.02	25	7	18	24	7	17	6.1490
2507.31	2507.32	13	3	10	12	3	9	8.9964	2513.22	2513.24	28	11	18	27	11	17	2.7614
2507.63	2507.58	13	1	12	12	1	11	9.9923		2513.25	26	8	19	25	8	18	5.1457
	2507.58	13	2	11	12	2	10	9.6413		2513.26	25	6	19	24	6	18	6.9898
	2507.70	13	7	8	14	7	7	5.6949	2513.39	2513.43	25	1	24	24	1	23	9.7267
2507.82	2507.81	14	2	13	13	2	12	10.0477	2513.52	2513.50	25	5	20	24	5	19	7.7821
	2507.84	14	3	12	13	3	11	9.3846		2513.52	26	7	20	25	7	19	5.9603
2508.10	2508.10	15	0	15	14	0	14	10.9160	2513.75	2513.75	27	8	19	26	8	18	4.9730
	2508.12	15	5	10	14	5	9	7.8759		2513.76	27	0	27	26	0	26	9.6717
2508.26	2508.25	16	7	10	15	7	9	5.9974		2513.76	26	2	25	25	2	24	9.3617
	2508.29	15	4	11	14	4	10	8.8574	2514.20	2514.14	26	4	23	25	4	22	3.1933
2508.49	2508.48	16	6	11	15	6	10	7.0896		2514.19	25	3	22	24	3	21	8.9968
	2508.49	15	3	12	14	3	11	9.6827		2514.20	28	1	28	27	1	27	9.2828
2508.79	2508.77	15	2	13	14	2	12	10.2802		2514.24	28	8	21	27	8	20	4.7790
	2508.79	18	9	10	17	9	9	4.3525	2514.41	2514.43	29	9	20	28	9	20	3.8902
	2508.80	17	7	10	16	7	9	6.2313	2514.75	2514.73	29	8	21	28	8	20	4.5693
2508.93	2508.88	16	2	15	15	2	14	10.5654		2514.77	28	6	23	27	6	22	6.2344
	2508.96	16	3	14	15	3	13	9.9208	2515.03	2515.01	29	7	22	28	7	21	5.2625
2509.08	2509.09	17	0	17	16	0	16	11.3049		2515.02	28	3	26	27	3	25	8.0601
	2509.09	18	8	11	17	8	10	5.3526		2515.07	30	1	30	29	1	29	8.4488
2509.22	2509.18	20	11	10	19	11	9	2.8939	2515.20	2515.21	30	8	23	29	8	22	4.3473

TABLE I (Continued)

Line Position (in cm ⁻¹)		Quantum Numbers							Line Position (in cm ⁻¹)		Quantum Numbers						
Exptl.	Theor.	J'	K' ₁	K' ₂	J''	K'' ₁	K'' ₂	$\frac{I''n''n'}{C'}$	Exptl.	Theor.	J'	K' ₁	K' ₂	J''	K'' ₁	K'' ₂	$\frac{I''n''n'}{C'}$
2515.47	2515.23	27	3	24	26	3	23	8.2831	2520.50	2520.37	39	3	36	38	3	35	3.5074
	2515.49	31	0	31	30	0	30	9.0127		2520.48	40	5	36	39	5	35	2.8955
	2515.49	30	7	24	29	7	23	4.9996		2520.53	41	7	34	40	7	33	2.1398
2515.61	2515.58	37	13	24	36	13	23	1.0562	2520.65	2520.61	43	1	42	42	1	41	2.8392
	2515.63	33	8	25	32	8	24	3.6409		2520.62	42	8	35	41	8	34	1.7014
2515.94	2515.91	32	1	32	31	1	31	7.5696		2520.69	42	3	40	41	3	39	2.8311
	2515.92	30	3	28	29	3	27	7.2690	2520.79	2520.76	45	2	43	44	2	42	2.0665
	2515.97	29	4	25	28	4	24	7.1072		2520.77	44	7	38	43	7	37	1.5617
	2515.97	30	5	26	29	5	25	6.2364		2520.81	45	9	36	44	9	35	1.0642
	2515.97	31	7	24	30	7	23	4.7280	2520.93	2520.91	45	0	45	44	0	44	2.6024
2516.19	2516.16	32	8	25	31	8	24	3.8802		2520.93	42	7	36	41	7	35	1.9342
	2516.23	29	3	26	28	3	25	7.4822		2520.94	41	6	35	40	6	34	2.3833
2516.56	2516.56	31	5	26	30	5	25	5.8759		2520.96	44	2	43	43	2	42	2.5632
2516.81	2516.78	32	3	30	31	29	6.4491		2521.08	2521.05	41	3	38	40	3	37	2.8860
	2516.80	33	1	32	32	1	31	6.5295		2521.08	42	4	39	41	4	38	2.5843
2517.00	2517.00	32	4	29	31	4	28	5.9983		2521.08	43	2	41	42	2	40	2.5588
	2517.03	31	4	27	30	4	26	6.3275		2521.11	44	9	36	43	9	35	1.1894
2517.13	2517.14	35	0	35	34	0	34	6.2399	2521.40	2521.39	44	3	42	43	3	41	2.3023
	2517.17	31	3	28	30	3	27	6.6385		2521.39	43	7	36	42	7	35	1.7404
2517.28	2517.24	35	9	26	34	9	25	2.7156		2521.45	44	8	37	43	8	36	1.3758
	2517.32	41	15	26	40	15	25	0.4526	2521.52	2521.51	45	9	36	44	9	35	1.0642
2517.40	2517.38	34	7	28	33	7	27	3.8935		2521.52	41	5	36	40	5	35	2.5863
	2517.42	33	2	31	32	2	30	6.0759	2521.65	2521.60	47	0	47	46	0	46	2.0979
2517.54	2517.54	36	1	36	35	1	35	5.8079		2521.65	46	2	45	45	2	44	2.0684
	2517.55	35	8	27	34	8	26	3.1643		2521.66	41	4	37	40	4	36	2.7355
	2517.56	31	5	26	30	5	25	5.8759		2521.71	43	3	40	42	3	39	2.3439
2517.73	2517.69	36	9	28	35	9	27	2.5180	2521.78	2521.76	45	2	43	44	2	42	2.0665
	2517.74	42	15	28	41	15	27	0.4151		2521.77	44	7	38	43	7	37	1.5617
	2517.19	37	10	27	36	10	26	1.9625		2521.81	44	4	41	43	4	40	2.0909
2517.79	2517.99	36	2	35	35	2	34	5.2775	2521.92	2521.91	46	9	38	45	9	37	0.9487
	2518.00	36	8	29	35	8	28	2.9312		2521.94	48	1	48	47	1	47	1.8744
2518.30	2518.28	39	11	28	38	11	27	1.3747		2521.94	47	10	37	46	10	36	0.7165
	2518.30	36	7	30	35	7	29	3.3486	2522.07	2522.06	44	5	40	43	5	39	1.9117
	2518.33	38	1	38	37	1	37	4.9780		2522.08	46	3	44	45	3	43	1.8474
2518.56	2518.57	36	6	33	35	6	30	3.7534	2522.25	2522.24	45	7	38	44	7	37	1.3941
	2518.57	38	9	30	37	9	29	2.1381		2522.27	49	0	49	48	0	48	1.6692
	2518.58	35	5	30	34	5	29	4.4509	2522.47	2522.42	47	2	45	46	2	44	1.6470
2518.72	2518.71	39	0	39	38	0	38	4.5844		2522.44	43	5	38	42	5	37	2.0890
	2518.73	36	4	33	35	4	32	4.4814		2522.50	46	4	43	45	4	42	1.6687
	2518.76	38	2	37	37	2	36	4.4985	2522.64	2522.61	50	1	50	49	1	49	1.4939
	2518.76	37	7	30	36	7	29	3.0862		2522.64	49	1	48	48	1	47	1.4750
2518.87	2518.88	35	3	32	34	3	31	4.9722		2522.67	47	8	39	46	8	38	0.9724
	2518.89	38	8	31	37	8	30	2.4844	2522.79	2522.74	48	3	46	47	3	45	1.4629
2519.04	2519.04	35	4	31	31	4	30	4.7544		2522.78	45	6	39	44	6	38	1.5436
	2519.09	40	1	40	39	1	39	4.2072		2522.80	46	5	42	45	5	41	1.5202
2519.21	2519.21	38	7	32	37	7	31	2.8332		2522.82	46	6	41	45	6	40	1.3828
	2519.21	38	3	36	37	3	35	4.1085	2522.97	2522.93	51	0	51	50	0	50	1.3221
2519.60	2519.58	37	5	32	36	5	31	3.7760		2523.97	47	3	44	46	3	43	1.4875
	2519.65	37	3	34	36	3	33	4.2057		2523.99	50	2	49	49	2	48	1.3058
	2519.65	38	5	34	37	5	33	3.4850	2523.09	2523.06	49	2	47	48	2	46	1.3066
2519.80	2519.76	41	9	32	40	9	31	1.6229		2523.06	48	8	41	47	8	40	0.8599
	2519.84	42	1	42	41	1	41	3.5070		2523.08	47	7	40	46	7	39	1.0998
2520.08	2520.08	40	7	34	39	7	33	2.3593		2523.10	45	4	41	44	4	40	1.7529
2520.21	2520.20	43	0	43	42	0	42	3.1857	2523.23	2523.24	52	1	52	51	1	51	1.1622
	2520.25	42	2	41	41	2	40	3.1342	2523.40	2523.38	50	3	48	49	3	47	1.1533
2520.34	2520.34	40	4	37	39	4	36	3.1500		2523.38	48	7	42	47	7	41	0.9734
	2520.34	43	10	33	42	10	32	1.1233		2523.42	52	11	42	51	11	41	0.3175
	2520.35	40	6	35	39	6	34	2.6360	2523.55	2523.50	48	5	44	47	5	43	1.1922

TABLE I (Continued)

Line Position (in cm^{-1})		Quantum Numbers							Line Position (in cm^{-1})		Quantum Numbers						
Exptl.	Theor.	J'	K' ₋₁	K' ₁	J''	K'' ₋₁	K'' ₁	$\frac{I''n''n'}{C'}$	Exptl.	Theor.	J'	K' ₋₁	K' ₁	J''	K'' ₋₁	K'' ₁	$\frac{I''n''n'}{C'}$
2523.55	53	0	53	52	0	52	1.0254		2523.71	2523.69	47	6	41	46	4	42	1.2134
2523.58	49	3	46	48	3	45	1.1726			2523.70	51	2	49	50	2	48	1.0149
2523.60	52	2	51	51	2	50	1.0137			2523.73	47	4	43	46	4	42	1.3754

$\Delta K_{-1} = 0, \pm 2, \dots$; and $\Delta K_1 = \pm 1, \pm 3, \dots$. Approximately 250 observed peaks, which are either individual or consist of a small number of closely spaced transitions, have been assigned. We have determined the band center for $\nu_1 + \nu_3$ to be $2499.60 \pm 0.10 \text{ cm}^{-1}$. This value may be compared with 2499.55 cm^{-1} reported by Shelton et al.,³ and 2499.0 cm^{-1} by Bailey and Cassia.¹⁰

Our results are potentially applicable to studies of the terrestrial atmosphere where SO_2 plays a serious role as a pollutant.¹¹ High-resolution infrared spectroscopy is a possible technique for the remote detection and monitoring of sulfur dioxide in situ. For example, solar spectra¹² in the 2500 cm^{-1} region may reveal SO_2 absorption features which would be susceptible to analysis. This spectral region is relatively free of absorption by H_2O and CO_2 , although CH_4 and N_2O may be significant there.⁷ Monochromatic laser emissions may be useful for studying terrestrial SO_2 in absorption. However, we have not yet been able to determine any relatively isolated and moderately strong lines in our laboratory spectra which fall close to observed laser oscillations,¹³ even if we include the effects of air-broadening.¹⁴

Finally, we note that $\nu_1 + \nu_3$ of SO_2 was recently considered¹⁵ in a determination of detectability limit of this gas in the atmosphere of the planet Mars. An upper limit of 0.0037 cm-atm was established,¹⁵ on the basis of the ν_3 fundamental, for SO_2 in the Martian atmosphere. This value may be compared with a terrestrial atmospheric value of the order of 1 cm-atm .¹⁶

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REFERENCES

1. K. Fox, G. D. T. Tejwani, and R. J. Corice, Jr., Research Report No. UTPA-ERAL-01, September, 1972.
2. R. J. Corice, Jr., K. Fox, and G. D. T. Tejwani, J. Chem. Phys., to be published January, 1973.
3. R. D. Shelton, A. H. Nielsen, and W. H. Fletcher, J. Chem. Phys. 21, 2178 (1953). References to earlier microwave and infrared measurements are given in this work.
4. J. Overend and H. W. Thompson, Proc. Roy. Soc. (London) A232, 291 (1955).
5. S. Saito, J. Mol. Spectry. 30, 1 (1969).
6. K. Fox, R. J. Corice, Jr., and G. D. T. Tejwani, to be published.
7. G. Herzberg, Infrared and Raman Spectra of Polyatomic Molecules (Van Nostrand, Princeton, N. J., 1945).
8. H. C. Allen, Jr. and P. C. Cross, Molecular Vib-Rotors (John Wiley and Sons, New York, 1963).
9. P. C. Cross, R. M. Hainer, and G. W. King, J. Chem. Phys. 12, 210 (1944).
10. C. R. Bailey and A. B. D. Cassie, Proc. Roy Soc. (London) A140, 605 (1933).
11. See, for example, W. W. Kellogg, R. D. Cadle, E. R. Allen, A. I. Lazrus, and E. A. Martell, Science 175, 587 (1972). An extensive bibliography on sulfur compounds in the atmosphere and oceans is given in this review article.
12. D. N. B. Hall, private communication, 1972.
13. W. S. C. Chang, Principles of Quantum Electronics (Addison-Wesley, Reading, Mass., 1969).
14. G. D. T. Tejwani, J. Chem. Phys., ~~to be published December,~~ 51, 4676 (1972).
15. D. Horn, J. M. McAfee, A. M. Winer, K. C. Herr, and G. C. Pimentel, Icarus 16, 543 (1972).
16. See, for example, C. W. Allen, Astrophysical Quantities (Oxford University Press, London and New York, 1963).